

Wave Packets

- Recall that for a wave packet $\Delta x \Delta k \sim 1$
- to localize a wave to some region Δx we need a spread of wavenumbers Δk
- de Broglie hypothesis $\lambda = h/p$ or $p = h/\lambda$
- but $k = 2\pi/\lambda$ and $p = (h/2\pi)(2\pi/\lambda) = \hbar k$
- \hbar bar
- $\Delta p = \hbar \Delta k$ and $\Delta x \Delta p \sim \hbar$
- $\Delta x \Delta p \geq \hbar$ uncertainty principle

Heisenberg's Uncertainty Principle

- States that measured values cannot be assigned to the position \mathbf{r} and momentum \mathbf{p} of a particle simultaneously with unlimited precision
- introduce "h-bar" $\hbar = h/2\pi$
- $\Delta x \Delta p_x \geq \hbar$
- $\Delta y \Delta p_y \geq \hbar$
- $\Delta z \Delta p_z \geq \hbar$

Energy-Time Uncertainty

- Recall for photons $E = hf$
- uncertainty in energy \Rightarrow uncertainty in f
- $\Delta f = \Delta E/h$ $\Delta\omega = 2\pi\Delta f$ $\Delta\omega \cdot \Delta t \approx 1$
- introduce “h-bar” $\hbar = h/2\pi$
- $\Delta E \cdot \Delta t \geq \hbar$

- need infinite time to make an accurate measurement of energy
- eg. Ground state of atom has well defined energy
- excited state does not

Uncertainty Principle

- If $\Delta p_x = \Delta p_y = \Delta p_z = 0$ (free particle)
- then $\Delta x = \Delta y = \Delta z = \infty$
- an electron has kinetic energy $K = 12.0 \text{ eV}$
- $K = (1/2)mv^2 = 12.0 \text{ eV} \Rightarrow v = 2.05 \times 10^6 \text{ m/s}$
- suppose the electron is moving // x and we can measure its speed with .50% precision (one part in 200)
- what is the uncertainty in its position?

- $p_x = mv = (.911 \times 10^{-31} \text{ kg})(2.05 \times 10^6 \text{ m/s})$
 $= 1.87 \times 10^{-24} \text{ kg.m/s}$
- $\Delta p_x = .005 (1.87 \times 10^{-24} \text{ kg.m/s})$
 $= 9.35 \times 10^{-27} \text{ kg.m/s}$
- $\Delta x \sim \hbar / \Delta p_x$
 $= (6.63 \times 10^{-34} \text{ J.s}) / (2\pi)(9.35 \times 10^{-27} \text{ kg.m/s})$
 $= 11.0 \text{ nm} \quad \sim 100 \text{ atomic diameters}$

Problem

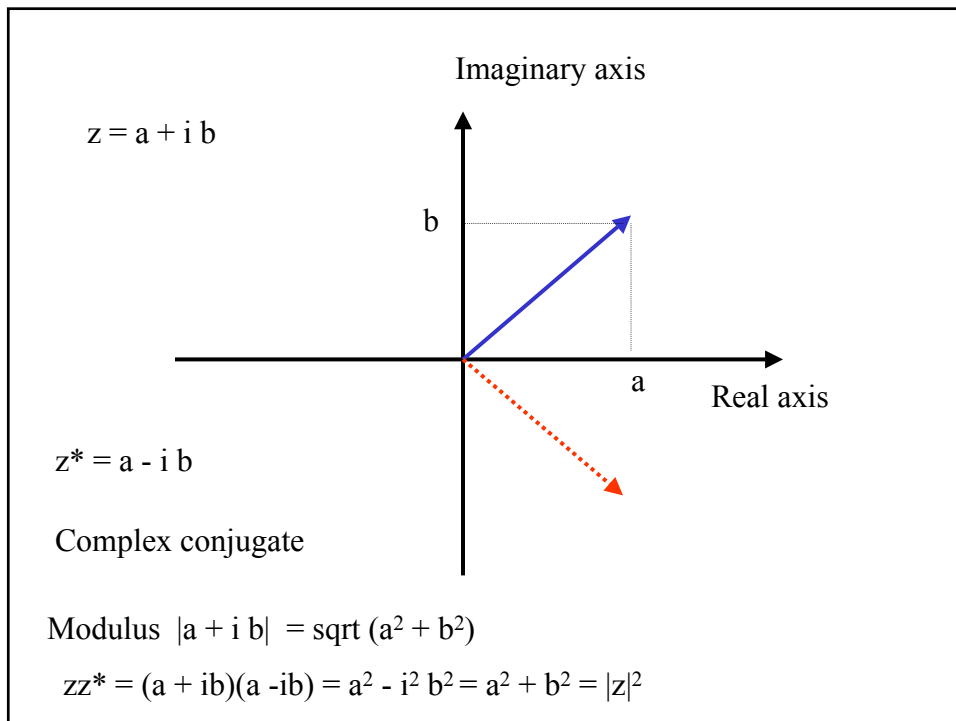
- Imagine a universe in which Planck's constant $\hbar = .60 \text{ J.s}$
- quantum baseball
- consider a .50 kg baseball that is moving at 20 m/s along an axis
- if the uncertainty in its speed is 0.5 m/s, what is the uncertainty in its position?
- $\Delta p = (.5 \text{ kg})(0.5 \text{ m/s}) = .25 \text{ kg.m/s}$
- $\Delta x \geq \hbar / (\Delta p) = (.60 / 2\pi) / .25 = .38 \text{ m}$

Wave function

- What kind of function is $\psi(x,y,z,t)$?
- Travelling wave? Standing wave?
- Let's learn something about **complex numbers**
- $\sqrt{-1}$ is a complex number $i^2 = -1$
- a complex number has the general form $a + i b$
- 'a' is the real part and 'b' is the imaginary part
- a **complex function** has the form $f_1(x) + i f_2(x)$

Real part

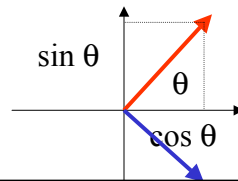
Imaginary part



Complex function

- $e^x = 1 + x + x^2/2 + x^3/6 + \dots + x^n/n! + \dots$
- $e^{i\theta} = 1 + i\theta + (i\theta)^2/2 + (i\theta)^3/6 + (i\theta)^4/24 + \dots$
- $= 1 + i\theta - \theta^2/2 - i\theta^3/6 + (\theta)^4/24 + \dots$
- $= (1 - \theta^2/2 + \theta^4/24 - \dots) + i(\theta - \theta^3/6 + \dots)$
- $= \cos \theta + i \sin \theta$
- $e^{-i\theta} = \cos \theta - i \sin \theta$

$$e^{i\theta} = \cos \theta + i \sin \theta$$



Complex Waves

- $\psi(x,t) = f(x) g(t)$ a standing wave
- string: $y(x,t) = 2y_m \sin(kx) \cos(\omega t)$
- matter wave: $\psi(x,t) = f(x) e^{-i\omega t}$
- $= f(x) [\cos \omega t - i \sin \omega t]$

Schrödinger's Quantum Mechanics

- Probability waves are described by a wave function $\psi(x,y,z,t)$
- Schrödinger adopted de Broglie's equations
- $\lambda = h/p$ wavelength of a particle
- $f = E/h$ frequency of a particle
- non-relativistic energy is
- $E = mc^2 + p^2/2m + V$
 - Rest energy K.E. P.E.

Quantum Mechanics

- Choose mc^2 as the reference energy
- $E = p^2/2m + V$ particle
- $hf = h^2/2m\lambda^2 + V$ wave
- relates f to λ
- what is the equation for the wave function $\psi(x,t)$?

Schrödinger Equation

$$\frac{d^2\Psi}{dx^2} + \frac{8\pi^2 m}{h^2} (E - V)\Psi = 0$$

- Schrödinger Equation 1926
- $\psi(x,t)$ is a solution of this equation
- the wave equation for matter waves

Free Particle

- Consider the case of $V=0$
- hence $E=(1/2)mv^2 = p^2/2m$

$$\frac{d^2\Psi}{dx^2} + \frac{8\pi^2 m}{h^2} (E - V)\Psi = 0$$

$$\frac{d^2\Psi}{dx^2} + \left(\frac{2\pi p}{h}\right)^2 \Psi = 0$$

Free Particle

$$\frac{d^2\Psi}{dx^2} + \left(\frac{2\pi p}{h}\right)^2 \Psi = 0$$

- But $p=h/\lambda$ and $k=2\pi/\lambda$

$$\frac{d^2\Psi}{dx^2} + k^2\Psi = 0$$

Can we solve this equation?